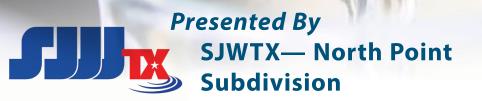
ANNUAL WATER OUALITY REPORTING VEAD 2020

REPORTING YEAR 2020



Este reporte incluye información importante sobre el agua para tomar. Para asistencia en español, favor de llamar al teléfono 830-312-4600.

From the General Manager

We are proud to provide you with our 2020 Consumer Confidence Report (CCR). The annual water quality report covers all testing performed between January 1 and December 31, 2020. Our team of professionals has spent countless hours collecting samples, analyzing data, focusing on superior quality water, and our vision, "to serve customers, communities, employees, shareholders and the environment at world-class levels." Our Mission, Vision, and Values bind us together to provide life-sustaining water for our customers, community and each other.

As you review the data in the Test Results section, keep in mind that many substances are detected at levels that vary throughout the year and at different locations. As a reminder, just because a substance is detected does not mean the water is unsafe. Natural waters, including the sources used by

SJWTX, contain a wide range of natural substances; in fact, some of the minerals detected are essential for good health.

The water source is one of the primary factors that affect the levels of the substances reflected in this report. SJWTX supplies both groundwater and surface water to our customers. Your system relies entirely on groundwater. Generally, groundwater is harder and contains more natural minerals than surface water. On the other hand, surface water typically contains small levels of natural organic substances and requires treatment by filtration. Regardless of the source, regulations require that we disinfect the water with chlorine and maintain a minimum level of chlorine residual throughout the distribution system.

Tip Top Tap

The most common signs that your faucet or sink is affecting the quality of your drinking water are discolored water, sink or faucet stains, a buildup of particles, unusual odors or tastes, and a reduced flow of water. The solutions to these problems may be in your hands.

Kitchen Sink and Drain

Hand washing, soap scum buildup, and the handling of raw meats and vegetables can contaminate your sink. Clogged drains can lead to unclean sinks and backed up water in which bacteria (i.e., pinkand black-colored slime growth) can grow and contaminate the sink area and faucet, causing a rotten egg odor. Disinfect and clean the sink and drain area regularly. Also, flush regularly with hot water.

Faucets, Screens, and Aerators

Chemicals and bacteria can splash and accumulate on the faucet screen and aerator, which are located on the tip of faucets, and can collect particles like sediment and minerals resulting in a decreased flow from the faucet. Clean and disinfect the aerators or screens on a regular basis.

Check with your plumber if you find particles in the faucet screen as they could be pieces of plastic from the hot water heater dip tube. Faucet gaskets can break down and cause black, oily slime. If you find this slime, replace the faucet gasket with a higher-quality product. White scaling or hard deposits on faucets and shower heads may be caused by hard water or water with high levels of calcium carbonate. Clean these fixtures with vinegar or use water softening to reduce the calcium carbonate levels for the hot water system.

Water Filtration/Treatment Devices

A smell of rotten eggs can be a sign of bacteria on the filters or in the treatment system. The system can also become clogged over time, so regular filter replacement is important. (Remember to replace your refrigerator filter!)

Quality First

A s in years past, we are committed to delivering the bestquality drinking water possible. To that end, we remain vigilant in meeting the challenges of new regulations, source water protection, water conservation, and



community outreach and education, while continuing to serve the needs of all our water users. Thank you for allowing us the opportunity to serve you and your family.

We encourage you to share your thoughts with us on the information contained in this report. After all, well-informed customers are our best allies.

QUESTIONS?

For more information about this report, or for any questions related to your drinking water, please contact Kristen Collier, Water Quality Specialist, at (830) 312-4600.



Contaminants in Source Water

To ensure that tap water is safe to drink, the U.S. EPA prescribes regulations limiting the amount of certain contaminants in water provided by public water systems. U.S. Food and Drug Administration regulations establish limits for contaminants in bottled water, which must provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of these contaminants does not necessarily indicate

that the water poses a health risk. The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it can acquire naturally occurring

minerals, in some cases, radioactive material; and substances resulting from the presence of animals or from human activity. Substances that may be present in source water include:

Microbial Contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, or wildlife;

Inorganic Contaminants, such as salts and metals, which can be naturally occurring or may result from urban storm-water runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming;

Pesticides and Herbicides, which may come from a variety of sources such as agriculture, urban storm-water runoff, and residential uses;

Organic Chemical Contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and which may also come from gas stations, urban storm-water runoff, and septic systems;

Radioactive Contaminants, which can be naturally occurring or may be the result of oil and gas production and mining activities.

Contaminants may be found in drinking water that may cause taste, color, or odor problems. These types of problems are not necessarily causes for health concerns. For more information on taste, odor, or color of drinking water, please contact our business office, (830) 312-4600. For more information about contaminants and potential health effects, call the U.S. EPA's Safe Drinking Water Hotline at (800) 426-4791.

Benefits of Chlorination

Disinfection, a chemical process used to control disease-causing microorganisms by killing or inactivating them, is unquestionably the most important step in drinking water treatment. By far, the most common method of disinfection in North America is chlorination.

Before communities began routinely treating

We remain vigilant in delivering the best-quality drinking water

99

66

drinking water with chlorine (starting with Chicago and Jersey City in 1908), cholera, typhoid fever, dysentery, and hepatitis A killed thousands of U.S. residents annually. Drinking water chlorination and filtration have helped to virtually eliminate these

diseases in the U.S. Significant strides in public health are directly linked to the adoption of drinking water chlorination. In fact, the filtration of drinking water plus the use of chlorine is probably the most significant public health advancement in human history.

How chlorination works:

Potent Germicide Reduction in the level of many disease-causing microorganisms in drinking water to almost immeasurable levels.

Taste and Odor Reduction of many disagreeable tastes and odors like foul-smelling algae secretions, sulfides, and odors from decaying vegetation.

Biological Growth Elimination of slime bacteria, molds, and algae that commonly grow in water supply reservoirs, on the walls of water mains, and in storage tanks.

Chemical Removal of hydrogen sulfide (which has a rotten egg odor), ammonia, and other nitrogenous compounds that have unpleasant tastes and hinder disinfection. It also helps to remove iron and manganese from raw water.

Important Health Information

You may be more vulnerable than the general population to certain microbial contaminants, such as *Cryptosporidium*, in drinking water. Infants, some elderly, or immunocompromised persons such as those undergoing chemotherapy for cancer; those who have undergone organ transplants; those who are undergoing treatment with steroids; and people with HIV/AIDS or other immune system disorders can be particularly at risk from infections. You should seek advice about drinking water from your physician or health-care provider. Additional guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* are available from the Safe Drinking Water Hotline at (800) 426-4791.



Source Water Assessment

TCEQ completed an assessment of your source water, and results indicate that some of our sources are susceptible to certain contaminants. The sampling requirements for your water system is based on this susceptibility and previous sample data. Any detections of these contaminants will be found in this Consumer Confidence Report. For more information on source water assessments and protection efforts at our system, contact Kristen Collier, Water Quality Specialist, at (830) 312-4600.

SYSTEM SUSCEPTIBILITY SUMMARY										
ASBESTOS	CYANIDE	METALS	MICROBIAL	MINERALS	RADIOCHEMICAL	SYNTHETIC ORGANIC CHEMICALS	DISINFECTION BYPRODUCT	VOLATILE ORGANIC CHEMICALS	DRINKING WATER CONTAMINANT CANDIDATE	OTHER
	MEDIUM	HIGH	MEDIUM	HIGH		HIGH	HIGH	HIGH	HIGH	

Where Does My Water Come From?

SJWTX North Point Subdivision provides ground water from the Trinity Aquifer.

SOURCE NAME / LOCATION	SOURCE WATER	TYPE OF WATER	REPORT STATUS	TCEQ SOURCE ID
North Point	Trinity Aquifer	Groundwater	Active	G0460235A G0460235B

Further details about sources and source-water assessments are available in Drinking Water Watch at the following URL: https://dww2.tceq.texas.gov/DWW/.

Water Main Flushing

Distribution mains (pipes) convey water to homes, businesses, and hydrants in your neighborhood. The water entering distribution mains is of very high quality; however, water quality can deteriorate in areas of the distribution mains over time. Water main flushing is the process of cleaning the interior of water distribution mains by sending a rapid flow of water through the mains.

Flushing maintains water quality in several ways. For example, flushing removes sediments like iron and manganese. Although iron and manganese do not pose health concerns, they can affect the taste, clarity, and color of the water. Additionally, sediments can shield microorganisms from the disinfecting power of chlorine, contributing to the growth of microorganisms within distribution mains. Flushing helps remove stale water and ensures the presence of fresh water with sufficient dissolved oxygen, disinfectant levels, and an acceptable taste and smell.

During flushing operations in your neighborhood, some short-term deterioration of water quality, though uncommon, is possible. You should avoid tap water for household uses at that time. If you do use the tap, allow your cold water to run for a few minutes at full velocity before use, and avoid using hot water to prevent sediment accumulation in your hot water tank.

Please contact us if you have any questions or if you would like more information on our water main flushing schedule.



Lead in Home Plumbing

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated

with service lines and home plumbing. This water supply is responsible for providing high-quality drinking water, but we cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline at (800) 426-4791 or at www.epa.gov/ safewater/lead.

Water Loss Audit

In the water loss audit submitted to the Texas Water Development Board during the year covered by this report, our system lost an estimated 348,400 gallons of water. If you have any questions about the water loss audit, please call (830) 312-4600.

Test Results

Our water is monitored for many different kinds of substances on a very strict sampling schedule. And, the water we deliver must meet specific health standards. Here, we only show those substances that were detected in our water (a complete list of all our analytical results is available upon request). Remember that detecting a substance does not mean the water is unsafe to drink; our goal is to keep all detects below their respective maximum allowed levels.

For each substance listed, compare the value in the Amount Detected column against the value in the MCL (or AL, SCL) column. If the Amount Detected value is smaller, your water meets the health and safety standards set for the substance. If there was a violation, you will see a detailed description of the event in this report.

The State recommends monitoring for certain substances less than once per year because the concentrations of these substances do not change frequently. In these cases, the most recent sample data are included, along with the year in which the sample was taken.

The Range column displays the lowest and highest sample readings. If the lowest sample reading and the highest sample reading are the same, that means that only a single sample was taken to test for the substance (assuming there is a reported value in the Amount Detected column).

If there is a 0, that means multiple samples were taken but the substance was not detected (i.e., below the detectable limits of the testing equipment).

If there is sufficient evidence to indicate from where the substance originates, it will be listed under Typical Source.

REGULATED SUBSTANCES

SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLED	MCL [MRDL]	MCLG [MRDLG]	HIGHEST AMOUNT DETECTED	RANGE LOW-HIGH	VIOLATION	TYPICAL SOURCE
Barium (ppm)	2020	2	2	0.0261	0.0261-0.0261	No	Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits
Beta/Photon Emitters ¹ (pCi/L)	2019	50	0	6.2	6.2–6.2	No	Decay of natural and man-made deposits
Chlorine ² (ppm)	2020	[4]	[4]	1.38	0.53-3.5	No	Water additive used to control microbes
Combined Radium (pCi/L)	2019	5	0	1.91	1.91-1.91	No	Erosion of natural deposits
Fluoride (ppm)	2019	4	4	1.23	1.23–1.23	No	Erosion of natural deposits; Water additive, which promotes strong teeth; Discharge from fertilizer and aluminum factories
Gross Alpha [excluding Radon and Uranium] (pCi/L)	2019	15	0	7	7–7	No	Erosion of natural deposits
Haloacetic Acids [HAAs] (ppb) ³	2019	60	NA	1.6	1.6–1.6	No	By-product of drinking water disinfection
Nitrate (ppm)	2020	10	10	0.09	0.09–0.09	No	Runoff from fertilizer use; Leaching from septic tanks, sewage; Erosion of natural deposits
TTHMs [Total Trihalomethanes] ³ (ppb)	2019	80	NA	6.1	6.1–6.1	No	By-product of drinking water disinfection
Uranium (ppb)	2019	30	0	1.8	1.8-1.8	No	Erosion of natural deposits

Definitions

90th %ile: The levels reported for lead and copper represent the 90th percentile of the total number of sites tested. The 90th percentile is equal to or greater than 90% of our lead and copper detections.

AL (Action Level): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements which a water system must follow.

MCL (Maximum Contaminant Level): The

highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.

MCLG (Maximum Contaminant Level

Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.

MRDL (Maximum Residual Disinfectant

Level): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

MRDLG (Maximum Residual Disinfectant Level Goal): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

NA: Not applicable.

ND (Not detected): Indicates that the substance was not found by laboratory analysis.

pCi/L (picocuries per liter): A measure of radioactivity.

ppb (parts per billion): One part substance per billion parts water (or micrograms per liter).

ppm (parts per million): One part substance per million parts water (or milligrams per liter).

SCL (Secondary Contaminant Level): These standards are developed to protect aesthetic qualities of drinking water and are not health based.

µmho/cm (micromhos per centimeter): A unit expressing the amount of electrical conductivity of a solution.

SUBSTANCE (UNIT OF MEASURE)	YEAR SAMPLE		AL	MCLG		t detected 'H %ile)		ABOVE AL/ AL SITES	VIOLATION	TYPICAL SOURCE	
Copper (ppm)	2019		1.3	1.3).066		0/5	No	Erosion of natural deposits; Leaching from wood prese systems.	rvatives; Corrosion of household plumbing
Lead (ppb)	2019)	15	0		0.5		0/5	No	Corrosion of household plumbing systems; Erosion of	natural deposits
SECONDARY SUBS	STANCES	5									
SUBSTANCE (UNIT OF MEASURE)		YEA SAMP		SCL	MCLG	HIGHEST AM DETECTE		RANGE LOW-HIGH	VIOLATION	TYPICAL SOURCE	
Chloride (ppm)		201	19	300	NA	21		21–21	No	Runoff/leaching from natural deposits	
Fluoride (ppm)		201	19	2.0	NA	1.23		1.23–1.23	No	Erosion of natural deposits; Water additive, which pro and aluminum factories	motes strong teeth; Discharge from fertilizer
Iron (ppb)		202	20	300	NA	29		29–29	No	Leaching from natural deposits; Industrial wastes	
Manganese (ppb)		202	20	50	NA	2.2		2.2–2.2	No	Leaching from natural deposits	
Sulfate (ppm)		201	19	300	NA	257		257–257	No	Runoff/leaching from natural deposits; Industrial wast	es
Total Dissolved Solid [TDS] (ppm)	s	201	19	1,000	NA	638		638–638	No	Runoff/leaching from natural deposits	
Zinc (ppm)		202	20	5	NA	0.289	0	0.289–0.28	9 No	Runoff/leaching from natural deposits; Industrial wast	es
UNREGULATED SU	JBSTAN	CES ⁴									¹ The MCL for beta particles is 4 mrem/year. The
SUBSTANCEYEAR(UNIT OF MEASURE)SAMPLE					EST AMO		NGE /-HIGH	TYPICAL	SOURCE		U.S. EPA considers 50 pCi/L to be the level of concern for beta particles. ² The highest amount detected is calculated as an average.
Bromodichloromethane (ppb) 2019			1.4	0	-1.4	Disinfec	tion by-produc	t			
Bromoform (ppb) 2019			1.5	0–1.5 Disin		Disinfec	tion by-produc	t	³ The value in the Highest Amount Detected		
Chloroform (ppb)			2019		1.2	0	-1.2	Disinfec	tion by-produc	t	column is the highest average of all sample results collected at a location over a year.
Dibromochloromethane (ppb) 2019		2019		2 1.2		.2–2	2 Disinfection by-product			⁴ Unregulated contaminants are those for	
Nickel (ppm) 2020		2020		0.0038	.0038 0.0038–0.		58 Discharge of drilling wastes; Discharge deposits		tes; Discharge from metal refineries; Erosion of natural	which EPA has not established drinking water standards. The purpose of unregulated contaminant monitoring is to assist EPA in	
Sodium (ppm)			2020		35	35–35 Erosic			of natural depo	sits	determining the occurrence of unregulated
OTHER UNREGUL	ATED SU	JBSTA	NCES	S ²							contaminants in drinking water and whether future regulation is warranted.
SUBSTANCE UNIT OF MEASURE)				YEAR SAMPLE		HEST AMOUN DETECTED		NGE -HIGH TY	PICAL SOURCE		
Bicarbonate (ppm)				2019		277	277-	–277 Ei	osion of natura	l deposits	
Calcium (ppm)				2020		79.3			osion of natura	-	
Dichloroacetic Acid (2019		1.6	1.6		isinfection by-p		
Diluted Conductance				2019		1110				rom natural deposits	
Gross Alpha [includin Uranium] (pCi/L)	ng Rador	n and		2019		8.1	8.1		rosion of natura		
Magnesium (ppm)				2020		48.4			rosion of natura	-	
Potassium (ppm)				2020		4.94			osion of natura	•	
Total Alkalinity [as C	[aCO3] (j	ppm)		2019		227	0-	227 Ei	osion of natura	l deposits	
		Total Hardness (ppm)		2020		397		–397 Ei	osion of natura		